

# Production of $^{38,39}\text{Cl}$ for Botany Tracer Studies

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# Botany and the Need for Chlorine Radiotracers

- Chlorine is an essential micronutrient for plants
- Often accumulates in soils to levels much higher than required for optimal plant growth
- Salt tolerance in plants has become a major research concern
- Little is known about the primary acquisition mechanisms of  $\text{Cl}^-$

# Mode of Production

- Proton irradiation of natural Argon gas
- Chlorine isotopes produced as ions and stick to target walls (similar to  $^{18}\text{F}$  –  $\text{F}_2$  production in  $^{18}\text{O}$ - $\text{O}_2$  gas targets)
- Chlorine ions washed off target walls with water

# Isotope Production By Proton Irradiation Of Argon

Reaction	Isotope	Half-Life	Decay Mode
$^{40}\text{Ar}(\text{p},3\text{n})$	$^{38}\text{K}$	7.63 m	$\beta^+$
$^{40}\text{Ar}(\text{p},2\text{pn})$	$^{38}\text{Cl}$	37.2 m	$\beta^-$
$^{40}\text{Ar}(\text{p},2\text{p})$	$^{39}\text{Cl}$	55.6 m	$\beta^-$

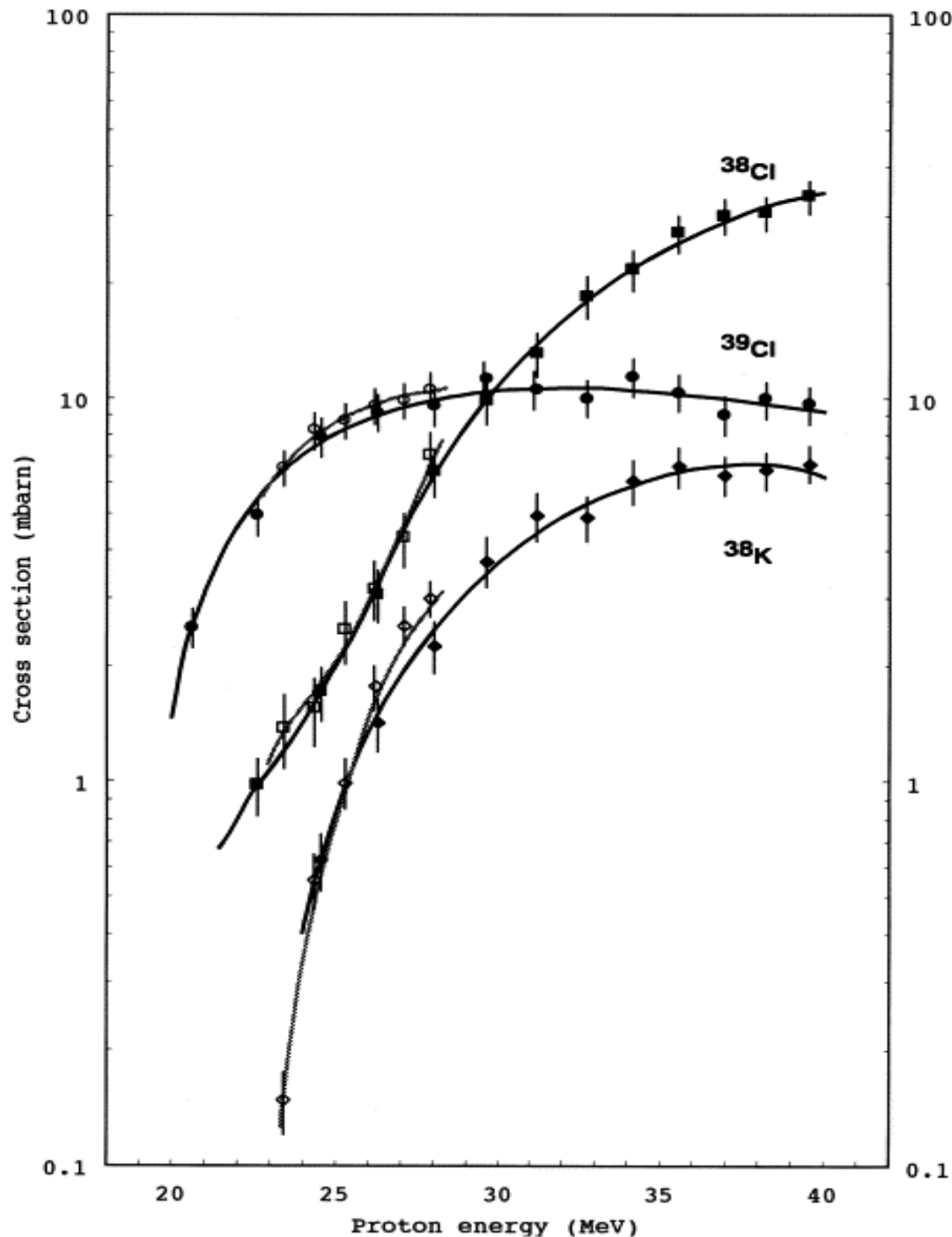


Figure 1.  
Excitation Functions of  
 $^{40}\text{Ar}(p,3n)^{38}\text{K}$ ,  $^{40}\text{Ar}(p,2pn)^{38}\text{Cl}$   
and  $^{40}\text{Ar}(p,2p)^{39}\text{Cl}$  reactions

From: K. Nagatsu, A. Kubodera, K. Suzuki.  
Excitation function measurements of  
 $^{40}\text{Ar}(p,3n)^{38}\text{K}$ ,  $^{40}\text{Ar}(p,2pn)^{38}\text{Cl}$  and  
 $^{40}\text{Ar}(p,2p)^{39}\text{Cl}$  reactions. Appl. Radiat. Isot.,  
50 (1999) 389-396

# $^{38,39}\text{Cl}$ Production Target

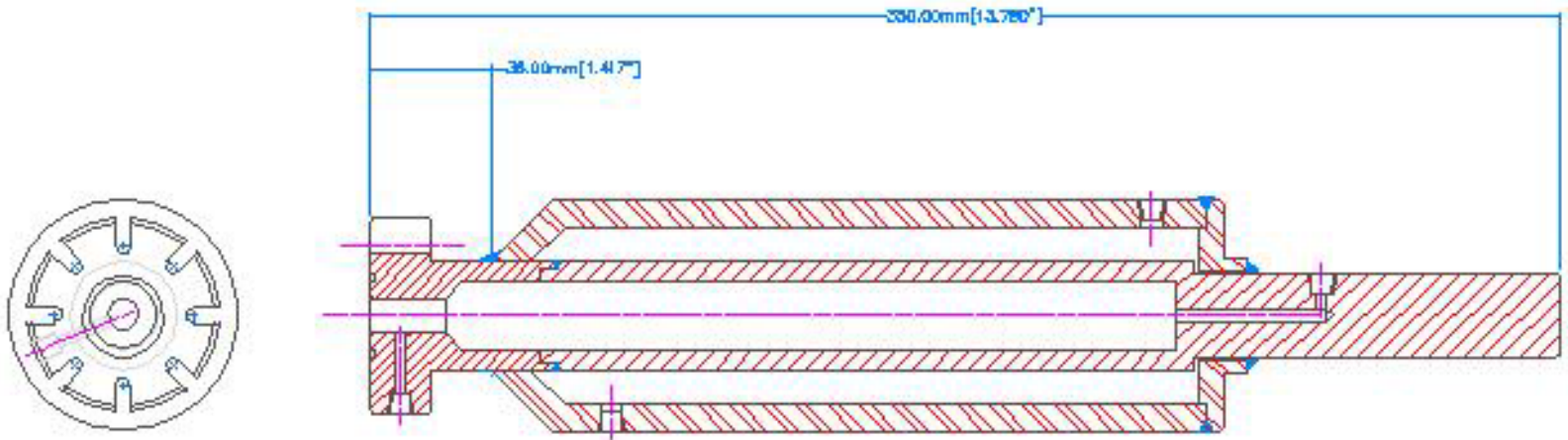
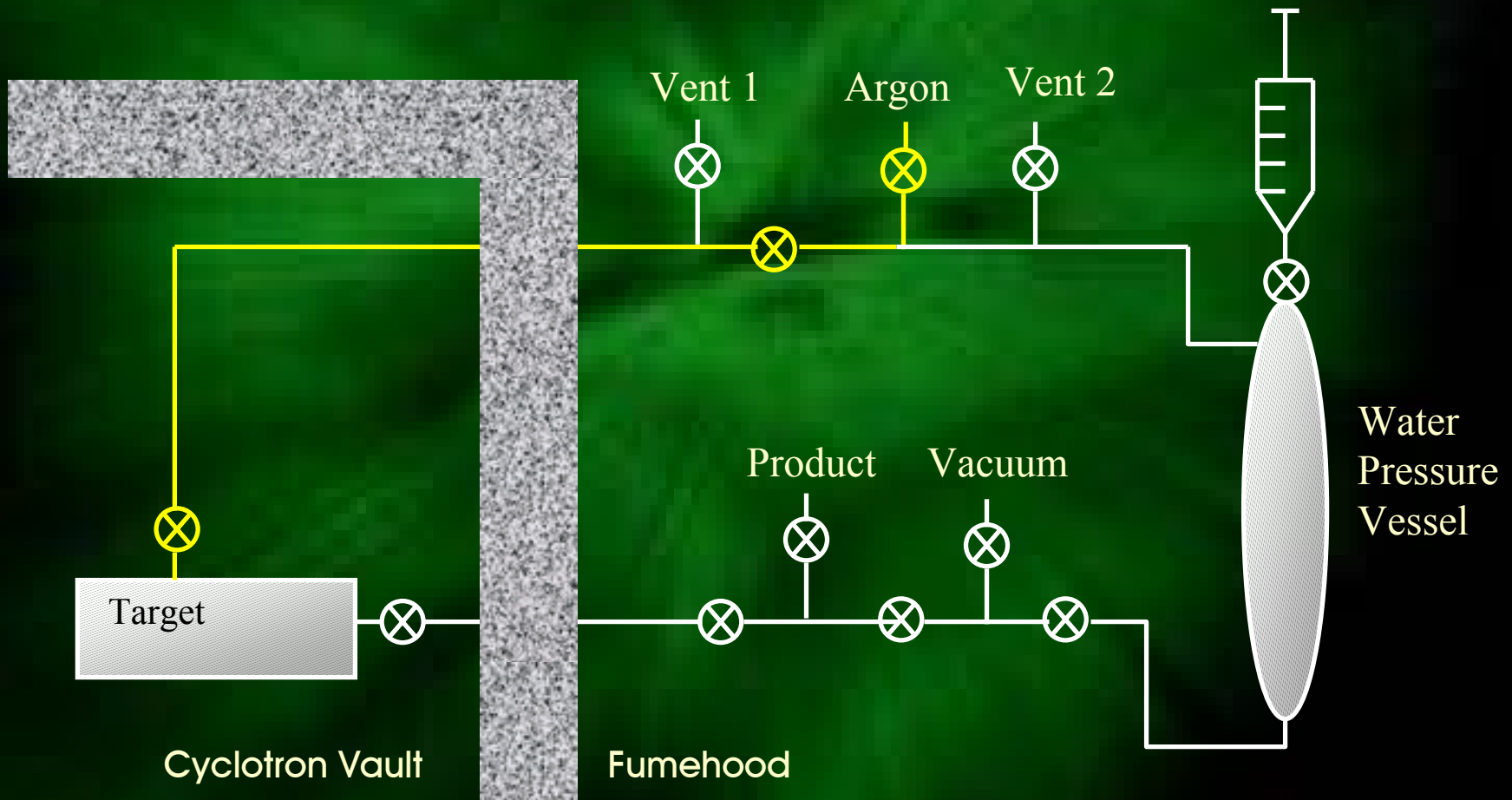
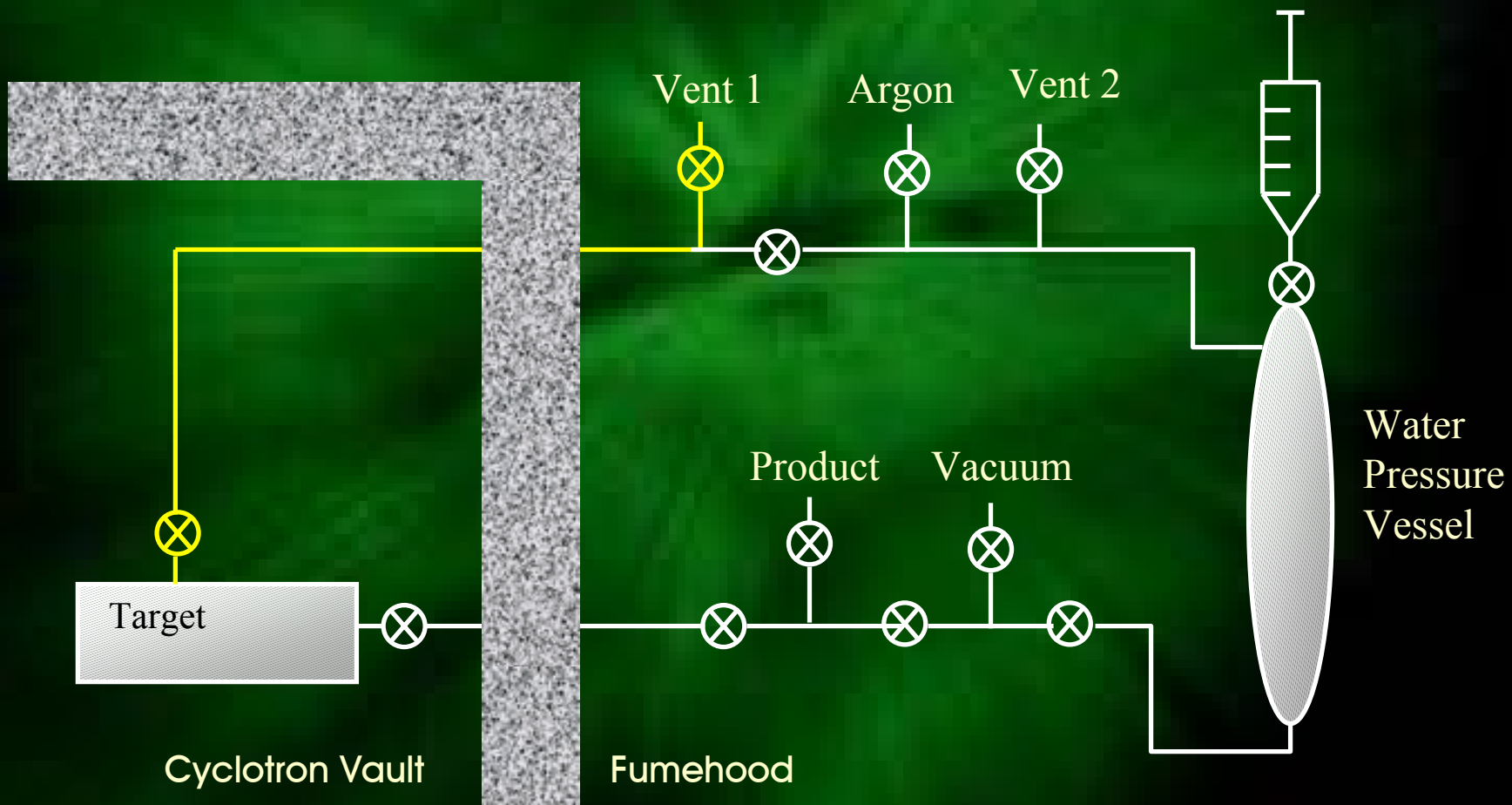


Figure 2. Aluminum Target Chamber

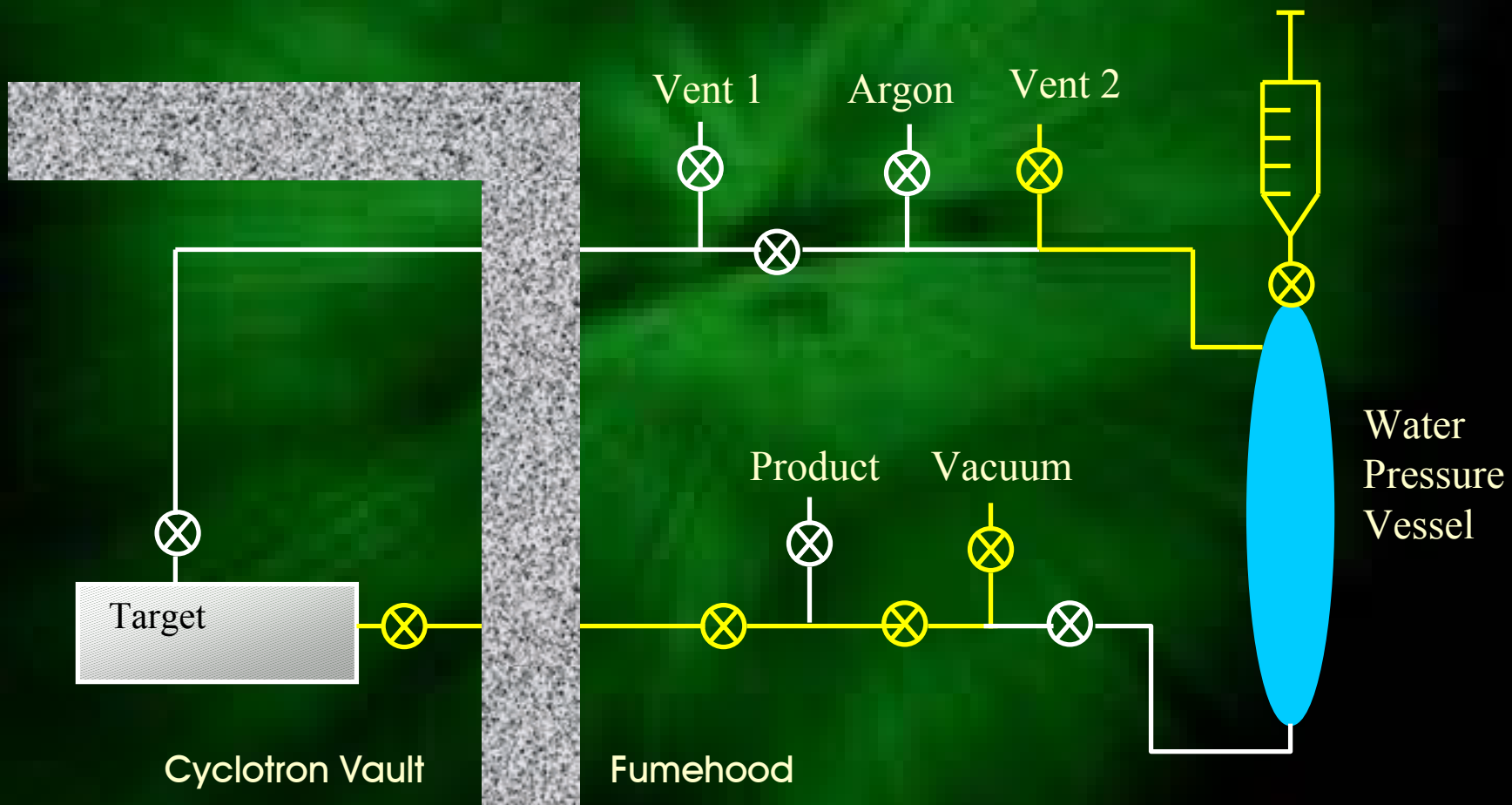
# $^{38,39}\text{Cl}$ Production System



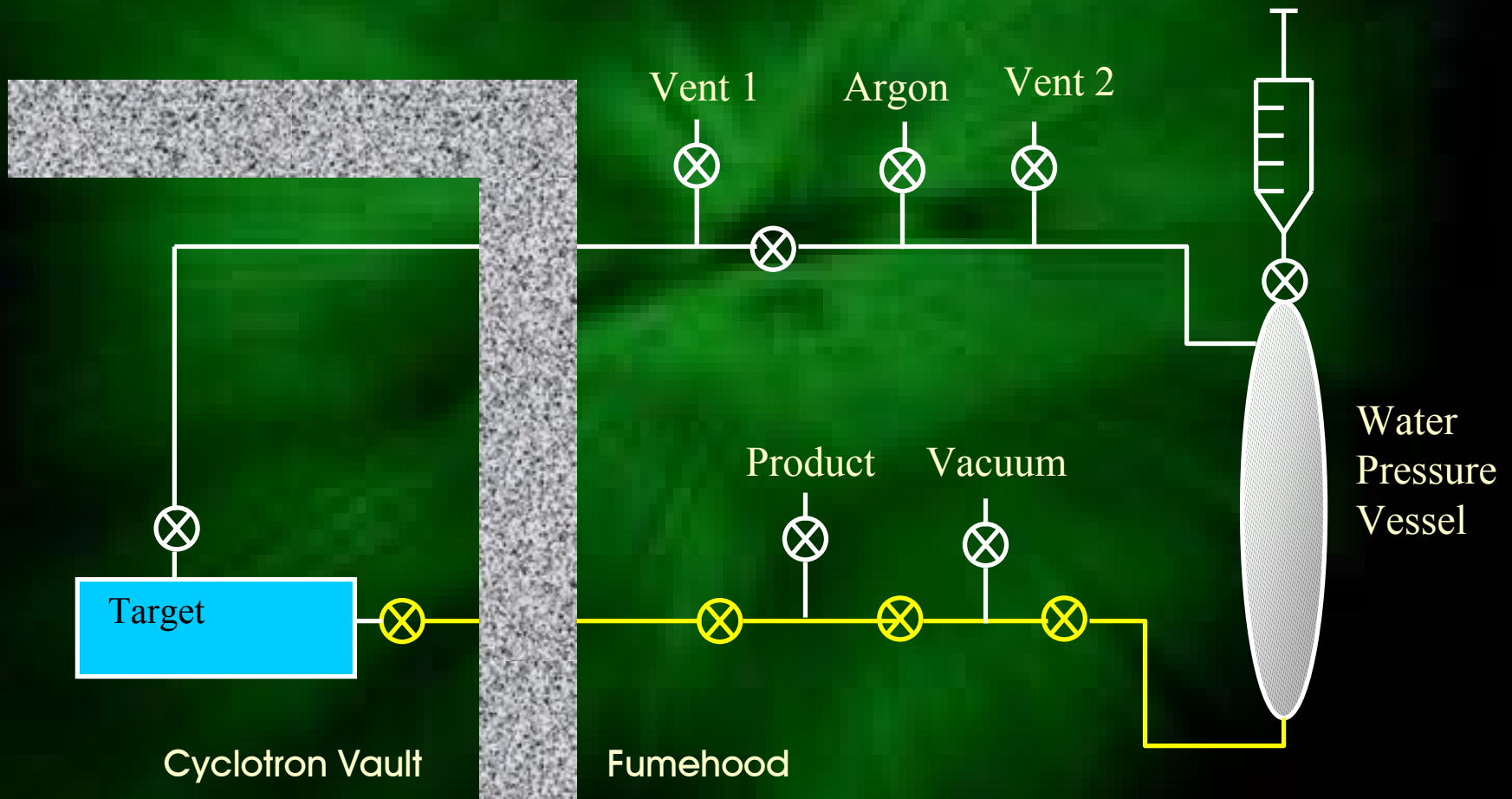
# $^{38,39}\text{Cl}$ Production System



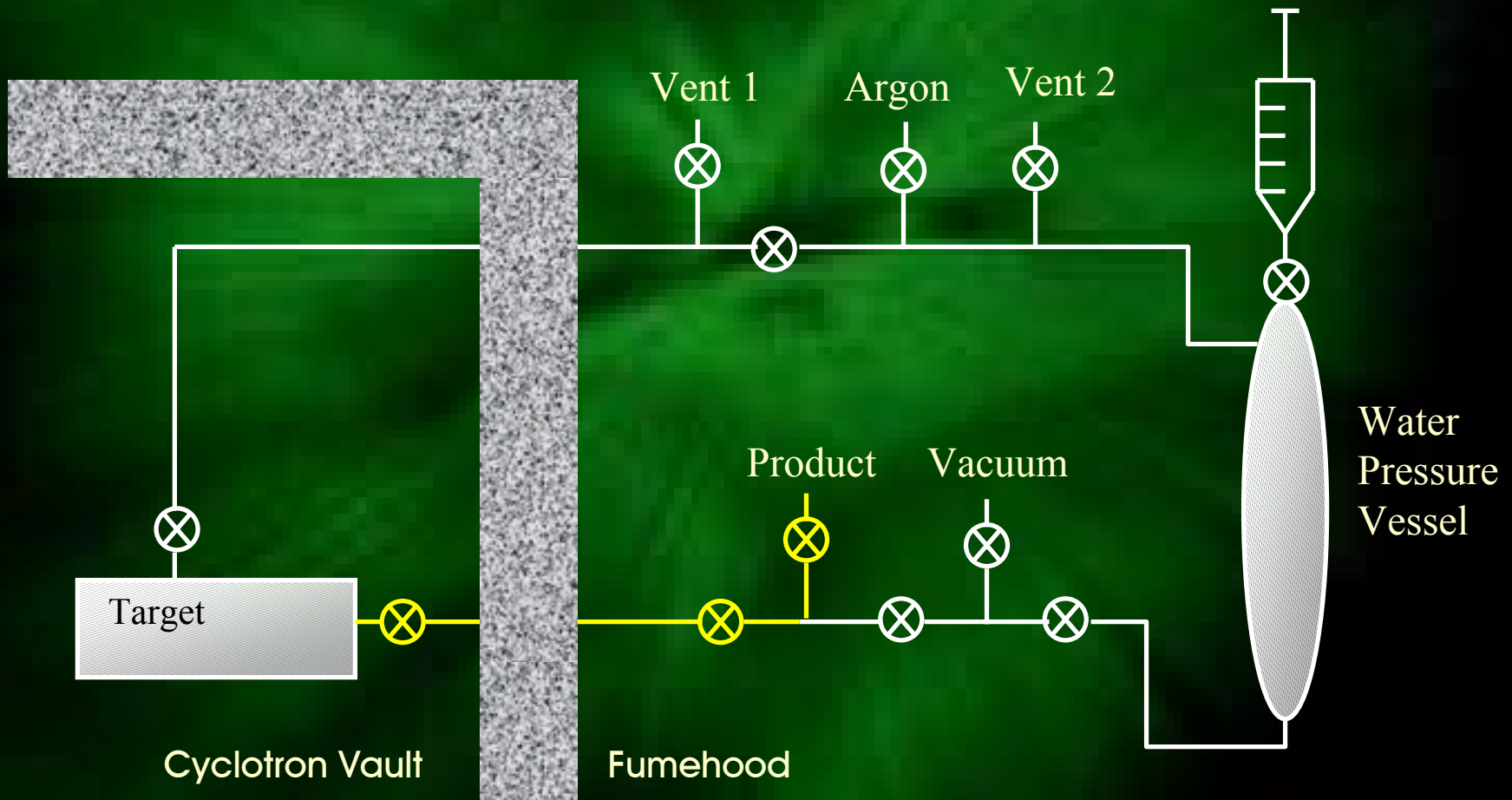
# $^{38,39}\text{Cl}$ Production System



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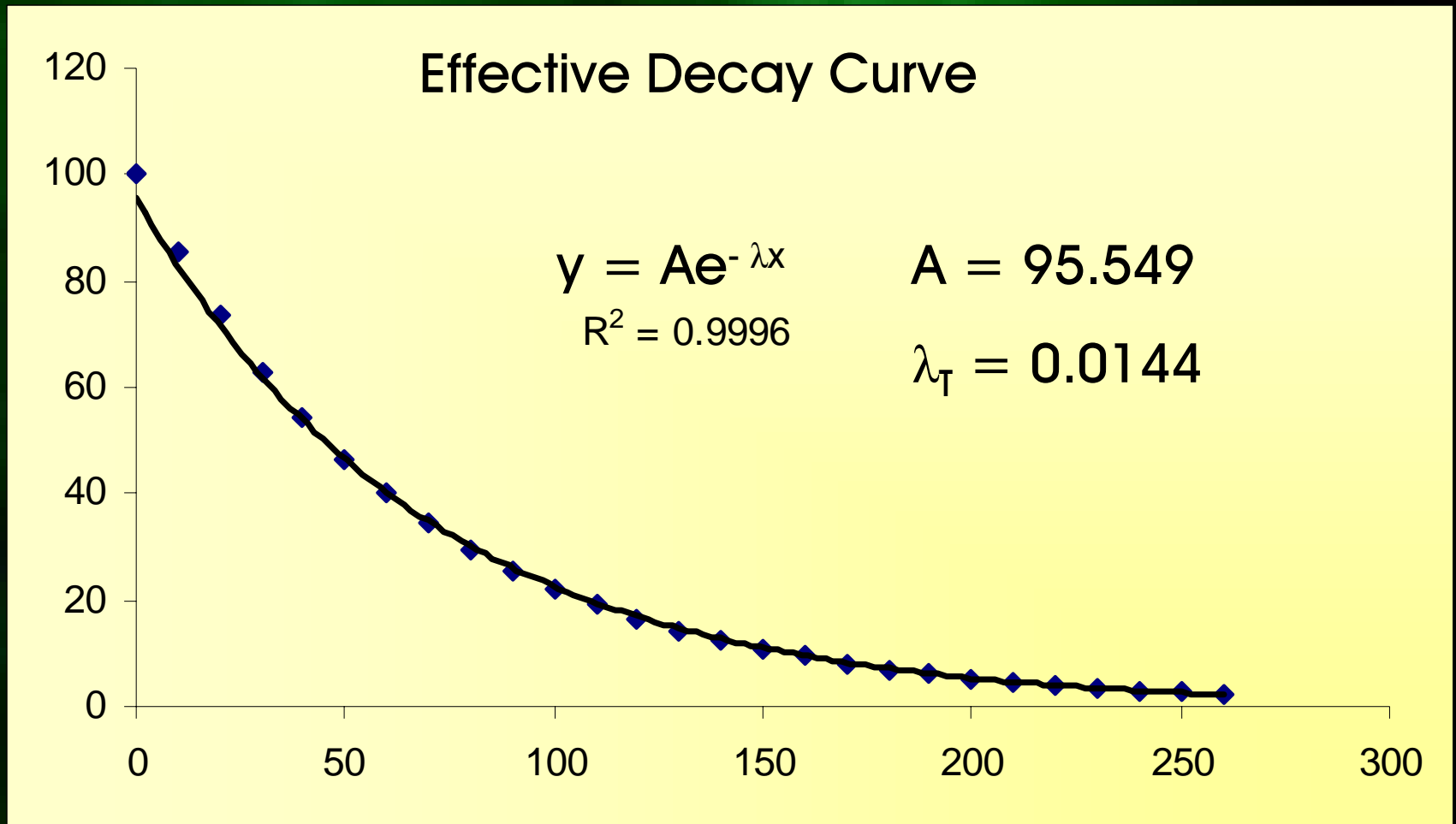
# Irradiation Parameters and Chlorine Recovery

- 250 psi Argon
- Irradiated with 5-7  $\mu\text{A}$  of 41 MeV protons for 30 minutes
- Target rinsed with very slightly basic solution
- Chlorine isotopes trapped on a Sepak anion exchange column
- Potassium isotopes passes through

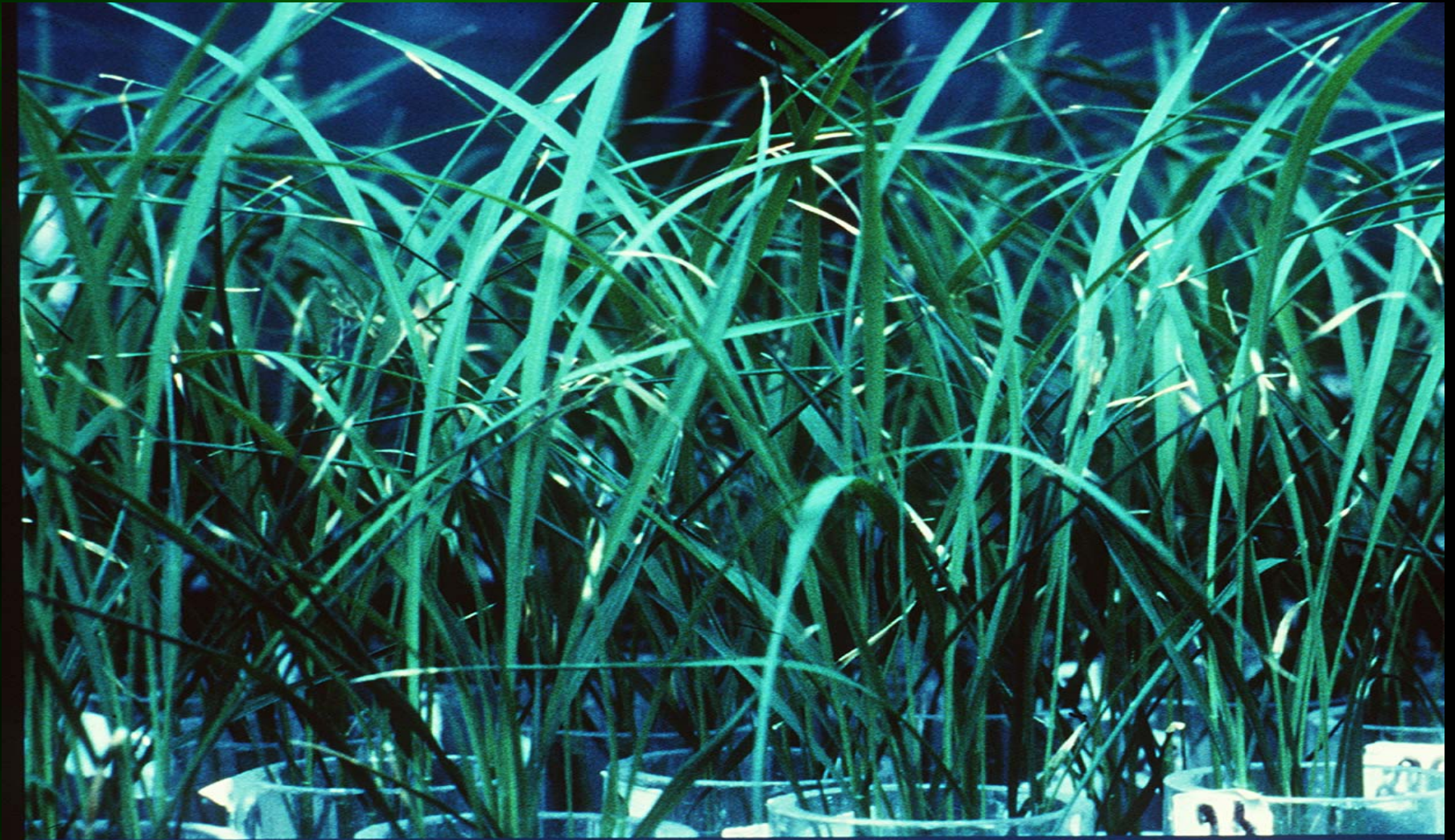
# Yields

- Typically produced 15-20 mCi at 10 minutes EOB (75%  $^{38}\text{Cl}$ , 25%  $^{39}\text{Cl}$ )
- 10% of theoretical recovery
- Losses probably from incomplete washing of the target chamber

# Effective Half-Life



# Botany Experiments



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- Cl isotopes were eluted from the exchange column with 20 ml of 10 mM  $\text{CaSO}_4$
- Barley seedling were transferred into the Cl loading solutions for a predetermined time to allow for uptake of the isotopes
- Efflux was measured by desorbing radioactivity from the roots into fresh solutions
- Radioactivity in roots, shoots and eluates was counted

# Results

- Using the tracers in the intact barley seedlings various transport systems were investigated using different salinity concentrations and nitrogen sources.
- First evidence of a saturable  $\text{Cl}^-$  low-affinity transport system responsible for translocation of  $\text{Cl}^-$  to the shoot

# Acknowledgements

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